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liquid water in particular as a structured liquid with an extensive hydrogen bonded structure the book examines water as a solvent for gases non electrolytes and electrolytes and reviews the properties sizes and thermodynamics of isolated and aqueous ions as well as their interactions including those of polyelectrolytes the effects of ions on water structure including those on solvent dynamics and certain thermodynamic quantities are presented this volume investigates water surfaces with its vapour with another liquid and with a solid as well as the effects of solutes including simple ions and the water miscible non electrolytes surfaces are relevant to biomolecular and colloidal systems and the book discusses briefly surfactants micelles and vesicles finally the book concludes with a review of the various biophysical implications involving chaotropic and kosmotropic ions in homogeneous solutions and the hofmeister series for ions concerning biomolecular and colloidal systems and some aspects of protein hydration and  $k_{\text{Na}}$  selectivity in ion channels ions in water and biophysical implications from chaos to cosmos will appeal to physical chemists biophysicists biochemists as well as to all students and researchers involved in the study of aqueous solutions

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